

Carbon 2 Chem®

L-O Analytical Challenges within the **Carbon2Chem®** Approach

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The successful on-line analysis of main and trace compounds in the ppb- and ppt-range contained in metallurgical gases before and after a purification system using PTR-TOF-MS has been demonstrated. However, not all trace compounds can be measured with this technique. Therefore, a complementary technique such as TD-GCMS is being applied to get the "whole" picture of traces, which could be harmful for the subsequent catalysts used for synthesis or even for substances breaking through and accumulating in the methanol product. Since off-line sampling methods are necessary for TD-GCMS analysis, the challenges associated with it have to be faced in order to validate the PTR-TOF-MS data.

Proton-Transfer-Reaction Time-of-Flight Mass Spectrometry



Add-ons: SRI and Fast-GC

 $H_3O^+(7,20 \text{ eV}): H_3O^+ + R \rightarrow RH^+ + H_2O$ NO⁺ (9,26 eV): $NO^+ + R \rightarrow R^+ + NO$ O_2^+ (12,07 eV): $O_2^+ + R \rightarrow R^+ + O_2^-$



Not all VOCs can be analyzed!



Figure 1 Sketch of the PTR-QiTOF-MS (+SRI) instrument with its advantages and drawbacks.



Challenges off-line sampling for TD-GCMS



Figure 4 Comparison chromatograms of the higher sensitivity of TD-GCMS towards trace analysis over conventional GCMS.

- Only a few traces detectable with GCMS, mainly by-products
- Over 150 traces in commercial methanols

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Retention time [min]

12

About 500 traces in raw C2C methanol

Method Validation: TD-GCMS & Valibration Gas Generator



Figure 3 Chromatograms of the different selectivity of TD-tubes towards trace compounds in raw metallurgical gases.

Figure 5 TD-GCMS (left) and calibration gas generator (right) used for method validation in Carbon2Chem[®]-Laboratory (Oberhausen).

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